

8.3 Acid Deposition

Question Paper

Course	DP IB Chemistry
Section	8. Acids & Bases
Topic	8.3 Acid Deposition
Difficulty	Medium

Time allowed: 20

Score: /10

Percentage: /100



Head to <u>savemyexams.co.uk</u> for more awesome resources

Question 1

TI II 6 I			" 1 6 .1 1 0
The pH of clean ra	ain water is around 5.5.	Which substance is res	oonsible for this?

- A. Methane
- B. Carbon dioxide
- C. Nitrogen oxides
- D. Sulfur dioxide

[1 mark]

Question 2

Which statement is **not** true about the increasing levels of atmospheric carbon dioxide?

$$CO_2(g) + H_2O(l) = H^+(aq) + HCO_3^-(aq)$$

$$HCO_3^-(aq) = H^+(aq) + CO_3^{2-}(aq)$$

- A. The concentration of HCO_3^- increases in water bodies
- B. The concentration of H⁺ increases in water bodies
- C. The pH of lakes and oceans falls as the CO_2 levels increase
- D. The increasing concentration of ${\rm CO_3}^{2-}$ enables marine organisms to build thicker shells

[1 mark]

Question 3

What is a major acid responsible for acid rain and originates from car engines?

- $A.H_2SO_4$
- $B.HNO_3$
- $C.SO_2$
- $D.NO_2$

[1 mark]

Question 4

What are three consequences of acid rain?

- I. Corrosion of marble statues
- II. Ozone depletion
- III. Leaching of Al³⁺
- A. I and II only
- B. I and III only
- C. II and III only
- $D.\,I,\,II\,and\,III$

[1 mark]

Question 5

Some coal contains as much as 3% sulfur. How much sulfur dioxide ($M = 64.07 \,\mathrm{g}\,\mathrm{mol}^{-1}$) is produced by burning a lump of coal of mass 125 g? ($A_r \,\mathrm{S} = 32.07$)

A.
$$\frac{32.07 \times 100 \times 64.07}{125 \times 3}$$

B.
$$\frac{125 \times 3 \times 64.07}{100 \times 32.07}$$

C.
$$\frac{125 \times 64.07}{32.07}$$

D.
$$\frac{125 \times 3 \times 32.07}{100 \times 64.07}$$

[1 mark]

Question 6

Which are true about the effect of acid deposition on plants?

- I. Essential minerals are lost
- II. Toxic minerals can be absorbed from the soil
- III. Respiration and photosynthesis are reduced
- A. I and II only
- B. I and III only
- C. II and III only
- D. I, II and III

[1 mark]

Question 7

Which are true about the effect of acid deposition on human health?

- I. Acid rain can burn the skin
- II. Acid particulates can trigger asthmatic attacks
- III. Toxic metal ions can be ingested
- A. I and II only
- B. I and III only
- C. II and III only
- D. I, II and III

[1 mark]

Question 8

The conversion of toxic emissions into harmless products takes place in catalytic converters as follows:

$$_{CO}(g) + _{NO}(g) \rightarrow _{CO_2}(aq) + _{N_2}(g)$$

The sum of the coefficients is

- A. 4
- B. 5
- C. 6
- D.7

[1 mark]



Head to <u>savemy exams.co.uk</u> for more awe some resources

Question 9

Which can	housodt	oroduco	NO	omissions	. 7
which can	be used to	oreduce	NO_{λ}	emissions	<i>:</i> :

- I. Catalytic converters in cars
- II. Lower temperature of combustion
- III. Blended gasoline-alcohol fuel
- A. I and II only
- B. I and III only
- C. II and III only
- $D.\,I,\,II\,and\,III$

[1 mark]

Question 10

Iron structures can be damaged by dry deposition such as the equation below:

$$Fe(s) + SO_2(g) + O_2(g) \rightarrow FeSO_4(s)$$

Which is true?

- I. Iron is a reducing agent
- II. Sulfur is a reducing agent
- III. The enthalpy change is ΔH_f
- A. I and II only
- B. I and III only
- C. II and III only
- D. I, II and III

[1 mark]