

2.2 Carbohydrates & Lipids

Question Paper

Course	DP IB Biology
Section	2. Molecular Biology
Topic	2.2 Carbohydrates & Lipids
Difficulty	Medium

Time allowed: 20

Score: /10

Percentage: /100

Two biological molecules are shown in **Diagram 1** below.

Diagram 1

Which row of the table correctly identifies features of these molecules?

	Molecule 1	Molecule 2
Α	Has 3 fatty acid chains	Fatty acid chains are all saturated
В	Contains 3 glycosidic bonds	Has 2 ester bonds and a phosphate group

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С	Has 3 saturated fatty acid chains	Has 1 unsaturated fatty acid chain	
D	Molecule is polar	Molecule is polar	

[1 mark]

Question 2

The molecular structure of starch makes it suited to its function. Which statement best explains why?

- A Many condensation reactions, in the breakdown of amylose and amylopectin, release stored energy.
- **B** Many hydrolysis reactions, in the formation of amylose and amylopectin, allow the release of stored energy to fuel cellular processes.
- **C** Amylose has a branched structure and amylopectin is coiled to give a compact structure for transport around the plant through the phloem.
- **D** The amylose-amylopectin complex is insoluble, so it does not affect the osmolarity of the cell.

There is a naturally-occurring polysaccharide which has the structure of an unbranched chain of the molecule acetylglucosamine held together by β -1,4 glycosidic bonds. Between these unbranched chains are many types of a much weaker bond. There are $-CH_2OH$ groups that alternate on each side of the polysaccharide chain.

Which of the following polysaccharides has a structure similar to that described above?

- A Glycogen
- **B** Cellulose
- **C** Amylopectin
- D Amylose

Which of the structures in **Diagram 2** correctly shows the structure of β -glucose and of α -glucose?

	β-glucose	α-glucose
A	CH₂OH OH OH OH	CH₂OH OH OH
В	CH ₂ OH OH OH	CH₂OH OH OH
С	CH ₂ OH OH OH OH	CH₂OH OH OH
D	CH ₂ OH OH OH	CH₂OH OH OH

Diagram 2

Which of the following statements correctly describes a feature of carbohydrates **OR** lipids?

- A Glycosidic bonds form during hydrolysis reactions, joining monosaccharides together to form disaccharides and polysaccharides.
- **B** A triglyceride is an example of a polymer as it is formed from many smaller, repeating subunits joined together by covalent bonds.
- **C** A triglyceride is not an example of a polymer although it is formed from smaller subunits joined together.
- **D** Glycosidic bonds join disaccharides together to form monosaccharides and polysaccharides.

[1 mark]

Question 6

Which of the following occurs when sucrose is formed from monosaccharides?

- A Condensation of glucose and fructose, using water.
- **B** Condensation of glucose and galactose, using water.
- **C** Condensation of glucose and fructose, releasing water.
- **D** Condensation of glucose and galactose, releasing water.

Which row of the table below contains two correct statements?

	Cis-fatty acids	Trans-fatty acids
А	Involves a saturated hydrocarbon chain	Involves an unsaturated hydrocarbon chain
В	H-atoms on the same side of a C=C double bond	H-atoms on different sides of a C=C double bond
С	Stack together more closely	Stack together further apart
D	Cause a kinked hydrocarbon chain	Cause a kinked hydrocarbon chain

[1 mark]

Question 8

Apart from being used for energy storage, lipids have a number of other roles. Which of the following is **not** a role of whole lipids?

- A Protection for soft internal organs.
- B Buoyancy aid.
- C Improving intestinal absorption of nutrients.
- **D** Regulators of gene expression.

Which of the following chemical formulae shows a carbohydrate molecule?

- A C₁₈H₃₄O₂
- **B** C₁₈H₃₂O₁₆
- C C₁₈H₃₂O₂
- **D** C₃H₈O₃

[1 mark]

Question 10

Which of the following is **not** a feature of lipids that contain trans-fatty acids?

- **A** They tend to form liquids at room temperature.
- **B** They increase the risk of coronary heart disease.
- **C** They are often labelled as 'partially hydrogenated vegetable oils' on food packaging.
- **D** They create more stable emulsions in food manufacture.