

# 8.3 Acid Deposition

## **Question Paper**

Course	DP IB Chemistry
Section	8. Acids & Bases
Торіс	8.3 Acid Deposition
Difficulty	Easy

Time allowed:	20
Score:	/10
Percentage:	/100

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Question 1	
Which gas is present in the largest proportion in the exhaust fumes of a car engine?	
A. carbon monoxide	
B. carbon dioxide	
C. water vapour	
D. nitrogen	
	[1 mark]
Question 2	
Oxides of nitrogen are released into the atmosphere in car engine exhaust fumes.	
Which environmental problem below is <b>not</b> made worse by these exhaust fumes?	
A. photochemical smog	
B. the hole in the ozone layer	
C. the corrosion of buildings	
D. acidification of lakes	
	[1mark]
Question 3 Which of the following is true?	
-	
A. Acid rain is caused by carbonic acid	
B. Acid rain turns universal indicator red	
C. Wet acid deposition includes snow and fog	
D. Catalytic converters remove oxides of sulfur from car exhausts	
	[1mark]



#### **Question 4**

What is the colour of universal indicator paper in rain water?

A. Red

- B. Blue
- C. Yellow
- D. Green

[1 mark]

#### **Question 5**

A sample of acid rain water is tested and found to be pH 4. How does its hydrogen ion concentration compare to pure water?

A. 1000 times more concentrated

- B.3 times more concentrated
- C.1000 times less concentrated
- D.3 times less concentrated

[1 mark]

#### **Question 6**

What happens to the oxidation state of sulfur when it becomes sulfuric acid in acid rain?

- A. It changes from 0 to -2
- B. It changes from 0 to +4
- C. It changes from 0 to +6
- D. It changes from -2 to +6

[1mark]



#### **Question 7**

One equation for the formation of acid rain from nitrogen dioxide is shown below.

 $2NO_2(g) + H_2O(I) \rightarrow HNO_2(aq) + HNO_3(aq)$ 

What type of reaction is this?

- A. Oxidation
- B. Reduction
- C.Redox
- D. Acid-base

[1 mark]

#### **Question 8**

Which of the following are used in catalytic converters?

- I. Ceramic
- II. Palladium
- III. Platinum
- A. I and II only
- B. I and III only
- C. II and III only
- D. I, II and III

#### [1 mark]

#### **Question 9**

A sample of rainwater is tested and found to be pH 6. Which of the following is **not** true?

- A. The sample is acidic
- B. The rainwater is acid rain
- C. Blue litmus would turn red in the sample
- D. The rain water is impure

[1mark]



### Question 10

Iron structures can be damaged by dry deposition such as the equation below:

 $Fe(s) + SO_2(g) + O_2(g) \rightarrow FeSO_4(s)$ 

Which is true?

- I. Iron is oxidised
- II. Sulfur is oxidised
- III. This is a neutralisation reaction
- A. I and II only
- B. I and III only
- C. II and III only
- D. I, II and III

[1 mark]