

10.1 Fundamentals of Organic Chemistry

Question Paper

Course	DPIB Chemistry
Section	10. Organic Chemistry
Topic	10.1 Fundamentals of Organic Chemistry
Difficulty	Hard

Time allowed: 20
Score: /10
Percentage: /100

Question 1

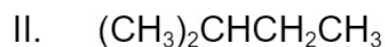
How many isomers, including structural and stereoisomers, with the formula C_5H_{10} have structures that involve π bonding?

A 3**B** 4**C** 5**D** 6

[1 mark]

Question 2

Study the formulae shown below and determine which molecules are isomers of each other

**A** I and II only**B** I and III only**C** II and III only**D** I, II and III

[1 mark]

Question 3

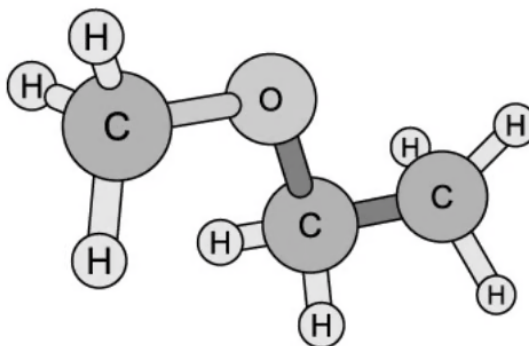
What is the correct condensed structural formula for 2,2-dibromo-4-methylhexane?

- A $\text{CH}_3\text{CBr}_2\text{CH}(\text{CH}_3)\text{CH}_2\text{CH}_2\text{CH}_3$
- B $\text{CH}_3\text{CHBrCBr}(\text{CH}_3)\text{CH}_2\text{CH}_2\text{CH}_3$
- C $\text{CH}_3\text{CBr}_2\text{CH}_2\text{CH}(\text{CH}_3)\text{CH}_2\text{CH}_3$
- D $\text{CH}_3\text{CHBrCH}(\text{CH}_3)\text{CHBrCH}_2\text{CH}_3$

[1 mark]

Question 4

What is the correct IUPAC name for the molecule shown?



- A ethoxyethane
- B methoxyethane
- C propanone
- D propanal

[1 mark]

Question 5

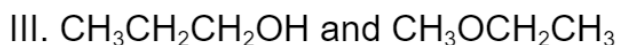
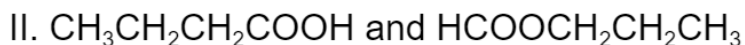
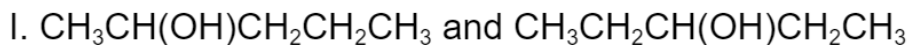
Which of the molecules shown below is **not** an isomer of pentan-2-ol?

- A pentan-1-ol
- B 2-methylbutan-2-ol
- C 2-methylpentan-2-ol
- D pentan-3-ol

[1 mark]

Question 6

Which of the following pairs are functional group isomers?

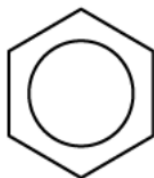


- A I and II only
- B I and III only
- C II and III only
- D I, II and III

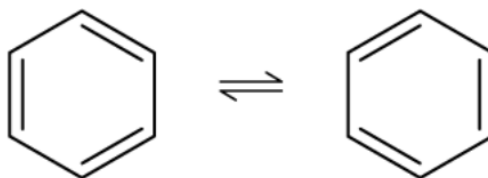
[1 mark]

Question 7

The structure of benzene is often shown as



This is a representation of a resonance hybrid structure that lies between these two possible structures



Evidence for this resonance structure is:

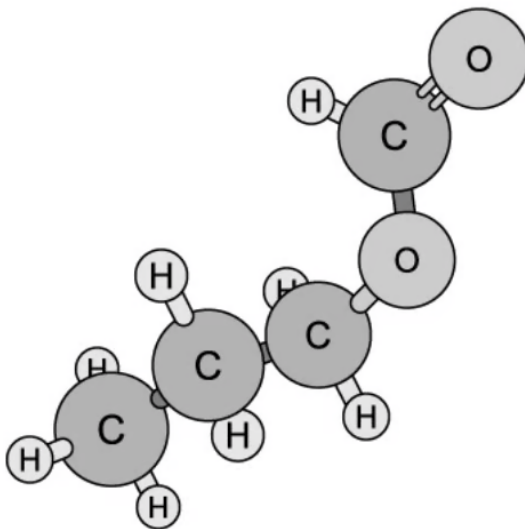
- I. The carbon-carbon bond lengths lie between the value for a single and a double bond
- II. The bond angles are all equal in benzene
- III. The enthalpy of hydrogenation of benzene is less exothermic than expected

- A I and II only
- B I and III only
- C II and III only
- D I, II and III

[1 mark]

Question 8

What is the correct name of the following molecule using IUPAC rules?

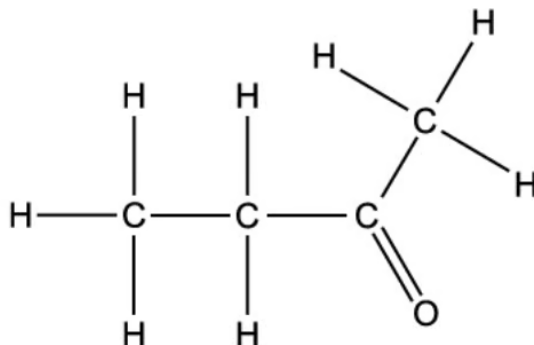


- A propyl methanoate
- B methyl propanoate
- C methoxypropane
- D butoxymethanal

[1 mark]

Question 9

What types of isomerism can the following molecule show?



- I. Branch-chain
- II. Positional
- III. Functional group

- A I and II only
- B I and III only
- C II and III only
- D I, II and III

[1 mark]

Question 10

Which row of the table is correct about the trend and explanation in the boiling points of the alcohols CH_3OH , $\text{C}_2\text{H}_5\text{OH}$ and $\text{C}_3\text{H}_7\text{OH}$?

	Trend in boiling points	Explanation
A	$\text{CH}_3\text{OH} > \text{C}_2\text{H}_5\text{OH} > \text{C}_3\text{H}_7\text{OH}$	The London dispersion forces decrease with each additional CH_2
B	$\text{CH}_3\text{OH} > \text{C}_2\text{H}_5\text{OH} > \text{C}_3\text{H}_7\text{OH}$	The strength of the hydrogen bonds decreases with each additional CH_2
C	$\text{C}_3\text{H}_7\text{OH} > \text{C}_2\text{H}_5\text{OH} > \text{CH}_3\text{OH}$	The London dispersion forces increase with each additional CH_2
D	$\text{C}_3\text{H}_7\text{OH} > \text{C}_2\text{H}_5\text{OH} > \text{CH}_3\text{OH}$	The strength of the hydrogen bonds increases with each additional CH_2

[1 mark]