

# 10.1 Fundamentals of Organic Chemistry Question Paper

Course	DP IB Chemistry
Section	10. Organic Chemistry
Topic	10.1 Fundamentals of Organic Chemistry
Difficulty	Medium

Time allowed: 20

Score: /10

Percentage: /100

I, II and III

D

Questi	on i										
How many structural isomers are there of tribromopropane, C <sub>3</sub> H <sub>5</sub> Br <sub>3</sub> ?											
Α	4		<b>B</b> 5		С	6		D	7		
											[1 mark]
Questi	on 2										
How many isomeric esters have the molecular formula $C_5H_{10}O_2$ ?											
Δ	4	В	5	c	9		<b>D</b> 7				
	7	J	O	J	Ü		<b>5</b> ,				
											[1 mark]
Questi	on 3										
Whic	ch fea	tures abou	ıt the n	nembers	of a ho	mologo	us serie	es are	corre	ct?	
	I. all members have the same general formula										
	II. the members have similar chemical properties										
	III. the members show a gradual change in physical properties										
Α	l a	and II only									
В											
С	II and III only										

Octane is a component of petrol or gasoline. Which of the following molecules could be an isomer of octane?

- A CH<sub>3</sub>CH<sub>2</sub>CHCHCHCH<sub>2</sub>(CH<sub>3</sub>)<sub>2</sub>
- $\mathbf{B} \qquad (\mathrm{CH_3})_2\mathrm{CHCH}(\mathrm{CH_2})_2(\mathrm{CH_3})_2$
- C CH<sub>3</sub>(CH<sub>2</sub>)<sub>5</sub>CH<sub>3</sub>
- D CH<sub>3</sub>CHCHCH(CH<sub>2</sub>)<sub>2</sub>(CH<sub>3</sub>)<sub>2</sub>

# Give the name of the following molecule

- A 2,4,4,4-trimethylethylbutane
- **B** 3,3,5-trimethylhexane
- C 2,4,4-trimethylhexane
- **D** 1,1,3,3-tetramethylpentane

How many branched-chain isomers are there which have the molecular formula  $C_5H_{12}$ ?

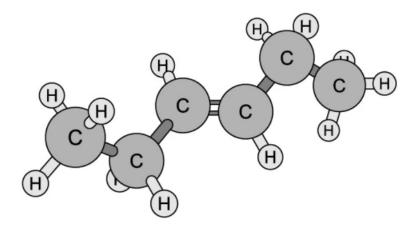
A 2

**B** 3

**C** 4

**D** 5

Study the diagram below and determine which of the three statements are correct



- I. A suitable name is hex-4-ene
- II. The empirical formula is CH<sub>2</sub>
- III. It is an isomer of cyclohexane
- A I and II only
- B I and III only
- C II and III only
- D I, II and III

Which of the molecules shown below contains an aldehyde functional group?

- A CH<sub>3</sub>CH<sub>2</sub>CH<sub>2</sub>OOCH<sub>3</sub>
- B CH<sub>3</sub>CH<sub>2</sub>CH<sub>2</sub>COCH<sub>3</sub>
- C CH<sub>3</sub>CH<sub>2</sub>OCH<sub>2</sub>CH<sub>3</sub>
- D CH<sub>3</sub>CH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>CHO

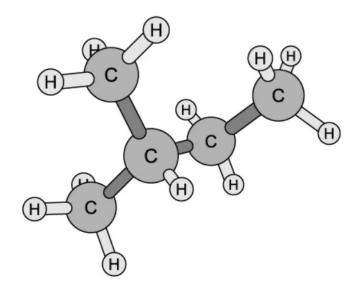
[1 mark]

#### Question 9

Which of the halogenoalkanes shown below contains a tertiary carbon atom?

- A CH<sub>3</sub>CH<sub>2</sub>CH<sub>2</sub>CI
- B CH<sub>3</sub>CH<sub>2</sub>CH(CH<sub>3</sub>)Br
- $\mathbf{C}$   $C(CH_3)_3CI$
- D CH<sub>3</sub>CHBrCH<sub>2</sub>CH<sub>3</sub>

What would be the name of the following compound using IUPAC rules?



- A pentane
- B ethylpropane
- C 2-methylbutane
- **D** 3-methylbutane